

Mo(CO)₆ monolayer adsorption and decomposition on Cu(111) studied by STM

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Valpré (Écully), 2011

Outline

Introduction

- **Mo-CO chemical bonds, decomposition; Applications in nanotechnology**

Mo(CO)₆ monolayer adsorption on Cu(111)

- **Structure interpretation, DFT calculation**

Mo(CO)₆ monolayer decomposition

Conclusions

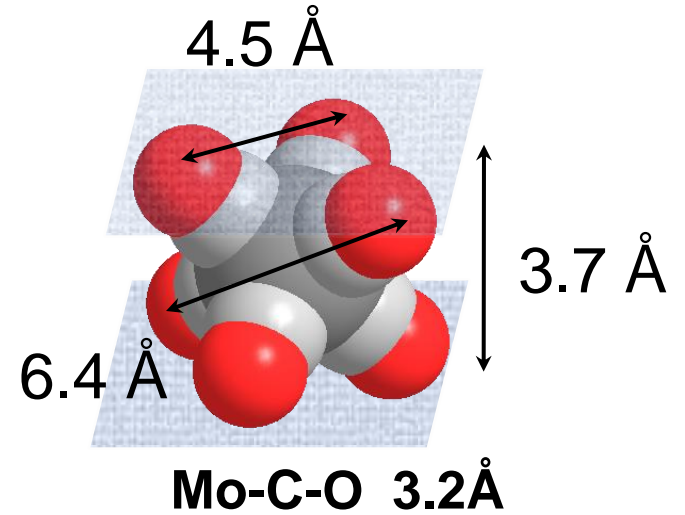
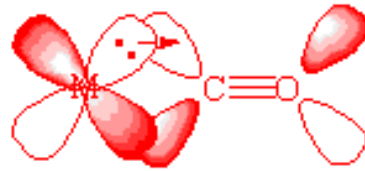
Mo(CO)₆ single molecule

σ bond:



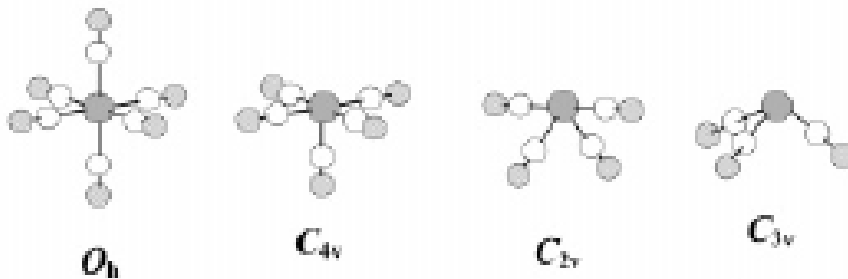
Neutral donor

π backbond:

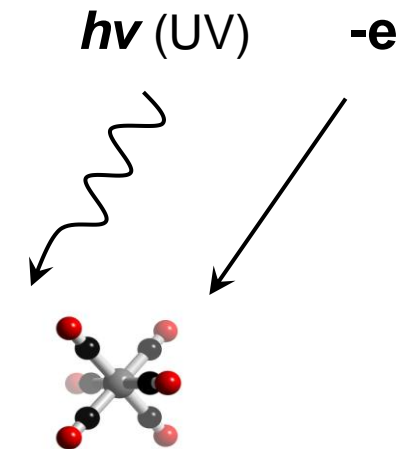


Calculated bond dissociation enthalpies on Mo(CO)_n ($n = 6, 5, 4,$ and 3)

≈ 1.4 eV/CO



Decomposition

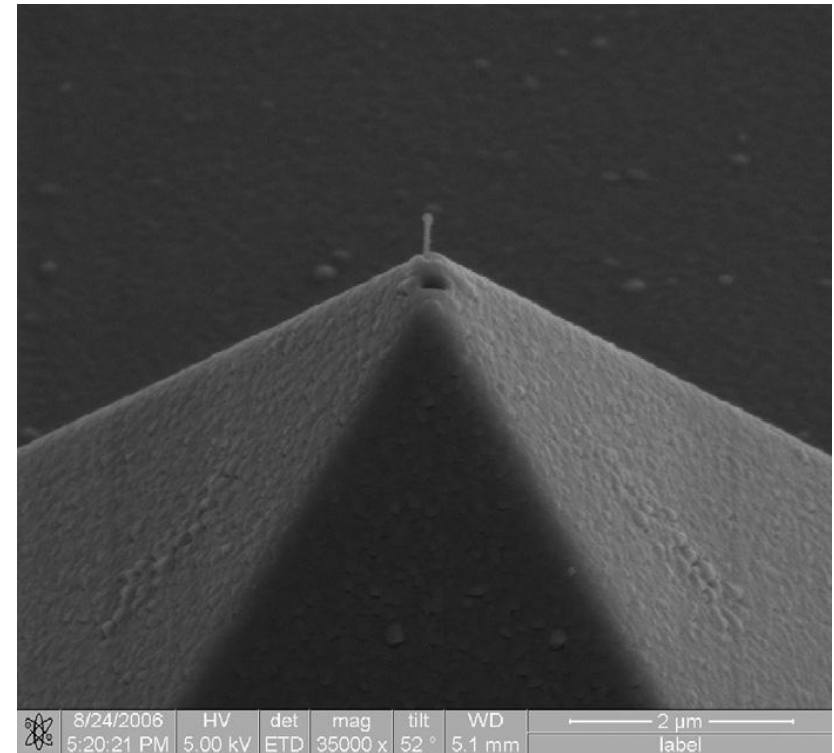


Y.Ishikawa, K.Kawakami,
J. Phys. Chem. A 111 (2007) 9940



Parameters for the electron beam-induced deposition process

Parameters	Experimental cond
Precursor gas	W(CO) ₆
Reservoir temperature	65°C
Capillary temperature	75°C
Base pressure	1.3×10^{-6} mbar
Process pressure (global)	4.6×10^{-6} mbar
Acceleration voltage	20kV
Beam current	257 pA
Dwell time per point	1000 µs
Repeat of e-beam irradiation	$N = 2, 4, 6$

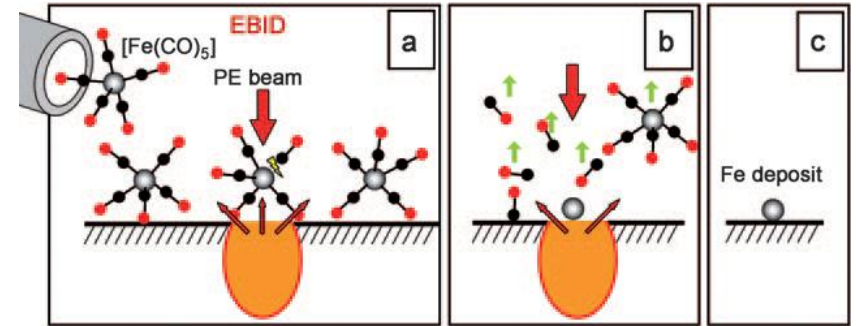
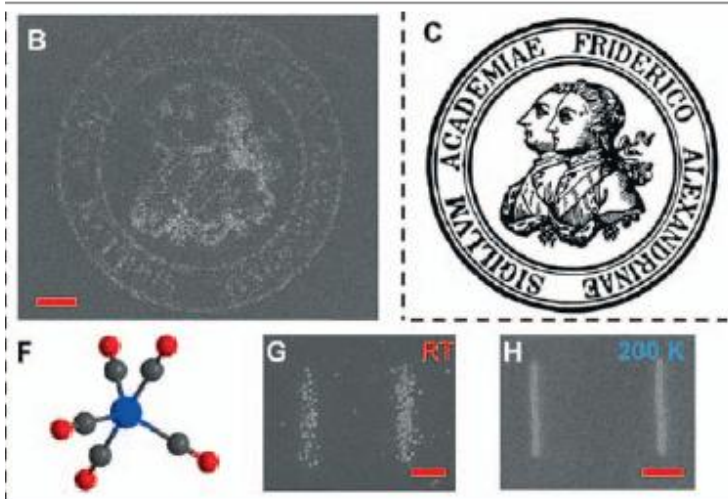


(AFM+SNOM)

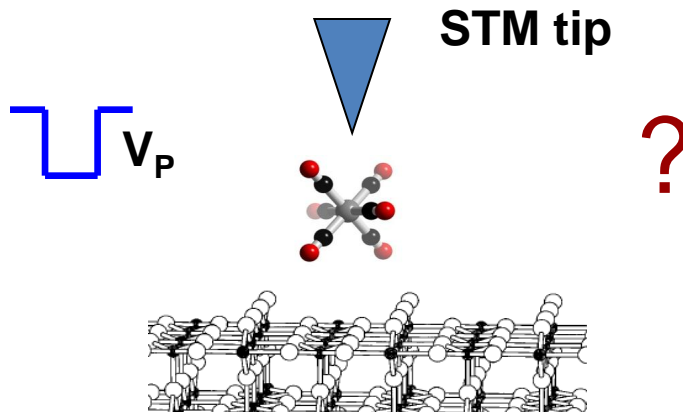
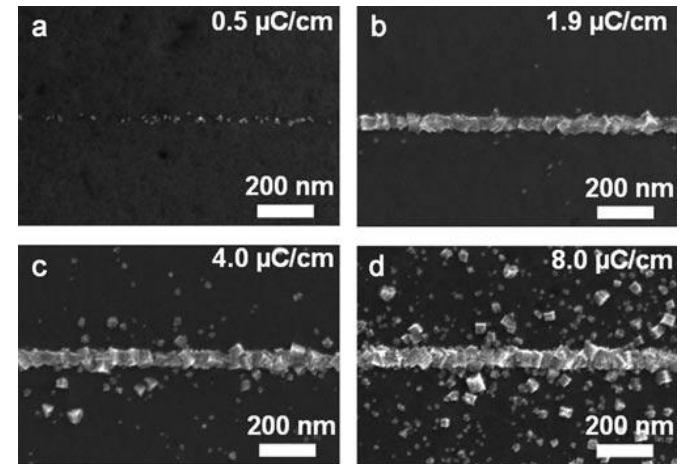
W.-S. Chang et al.

Ultramicroscopy 108 (2008) 1070–1075

Direct Writing with an Electron-Beam in UHV



T.Lukasczyk, M.Schirmer, M.-M.Walz, F.Vollnhals, H.-P.Steinrück and H. Marbach, *Omicron NanoTech. Newsletter*, Vol. 12 No. 2, 2008



Angew. Chem. Int. Ed. 2010, 49, 4669–4673

Experimental details

VT STM Omicron Nanotechnology

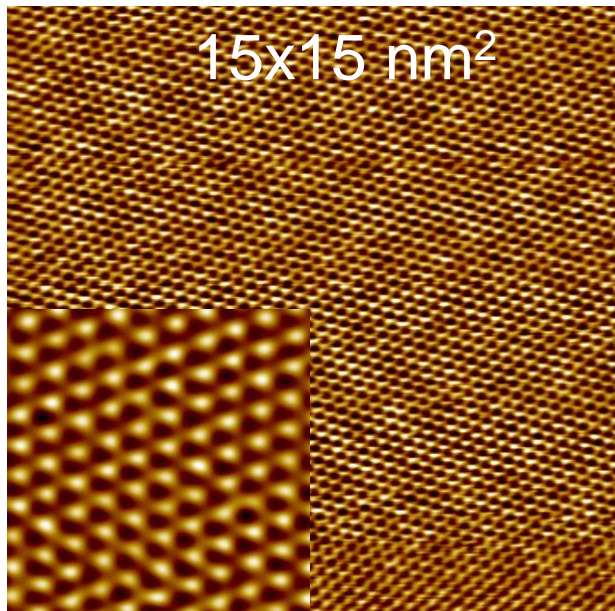
UHV base pressure 1×10^{-10} mbar

LN cooling system $150 \div 800$ K

$\text{Mo}(\text{CO})_6$ gas by

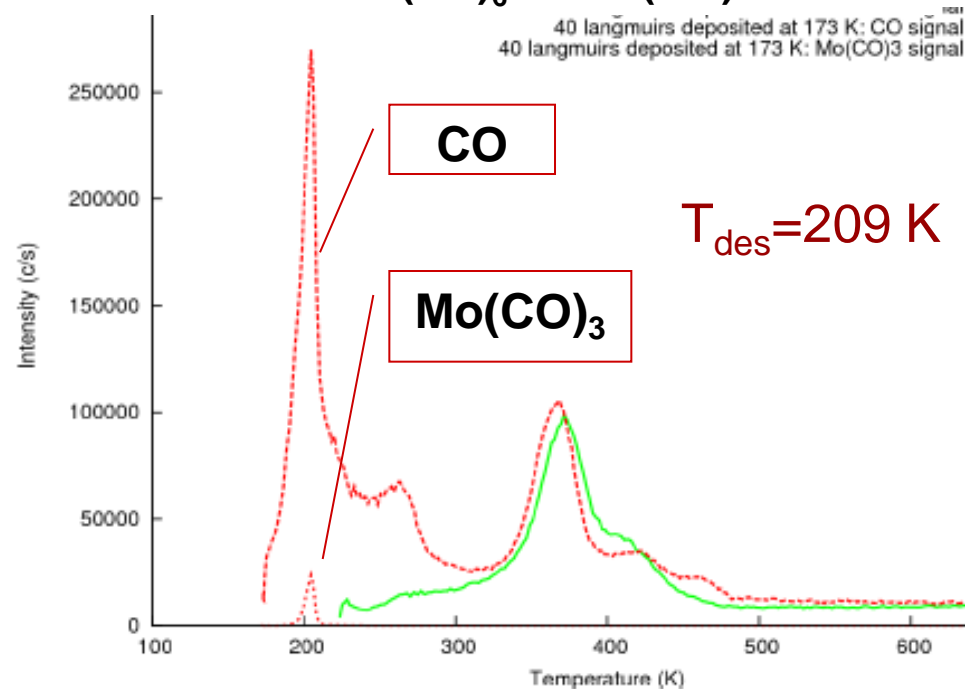
powder evaporation at RT

$\text{Cu}(111)$, $a=2.55 \text{ \AA}$



$2.2 \times 2.2 \text{ nm}^2$

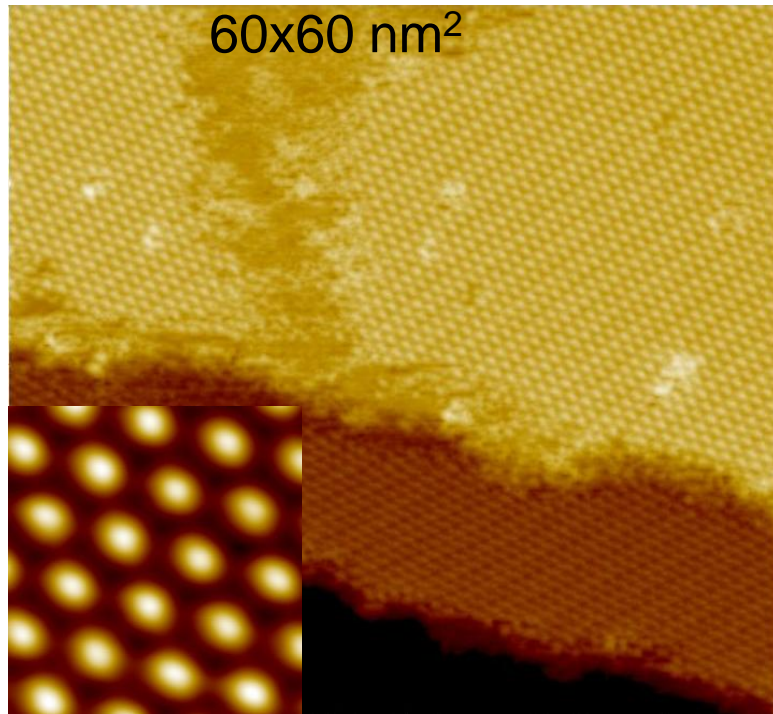
Temperature Programmed Desorption $\text{Mo}(\text{CO})_6$ on $\text{Cu}(111)$



$T = 160 \text{ K}$

Mo(CO)₆ monolayer on Cu(111)

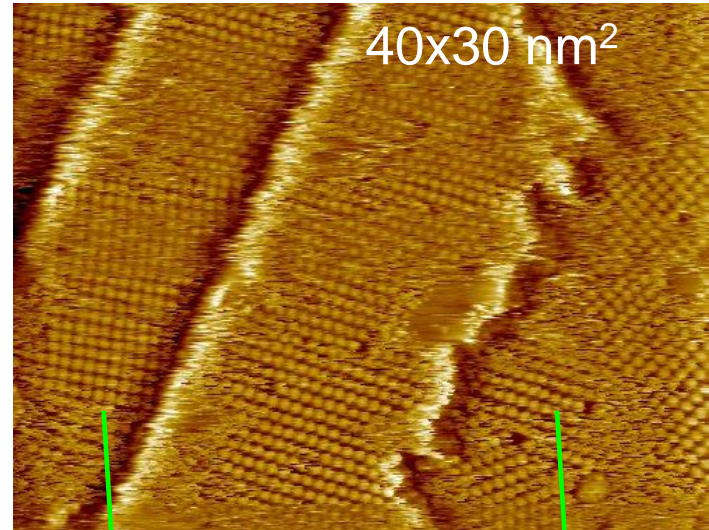
Exposition: 1-2 L



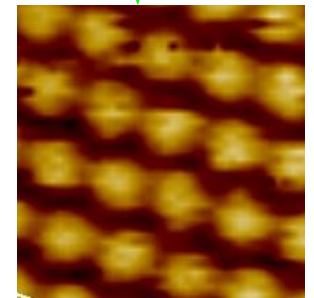
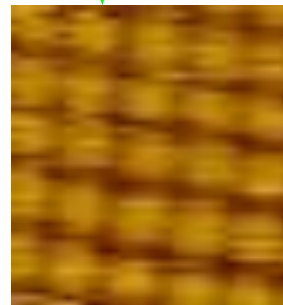
$I_t = 5 \text{ pA}; V_b = 20 \text{ mV}$

Apparent height $\sim 0.6 \div 0.8 \text{ \AA}$;
Corrugation of ML $\sim 0.4 \text{ \AA}$

5-10 L

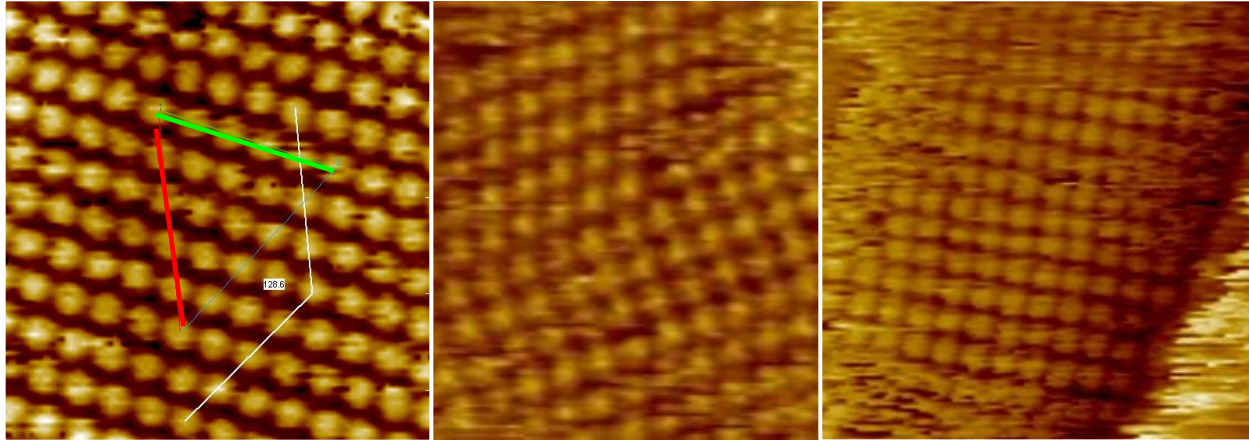


$I_t = 10 \text{ pA}; V_b = 200 \text{ mV}$



Cu(111) + CO interface

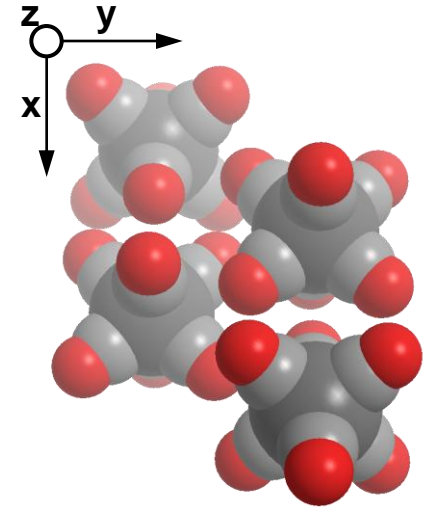
Mo(CO)₆ monolayer on Cu(111) (CO)



Hex-like : green 6.8 Å (±0.1)
red 7.4 Å (±0.1)

square-like : 6.8 Å (±0.1)

Bulk: Oh; P_{nma}

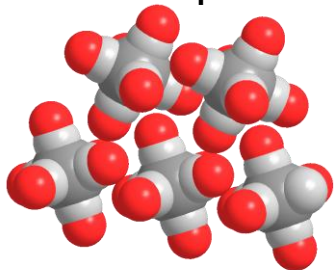


x=12.019; y=11.415;
z=6.488 Å; N=4.

[2-10]

a=7.20; b=6.49 Å; φ=63.2

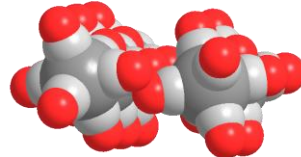
top



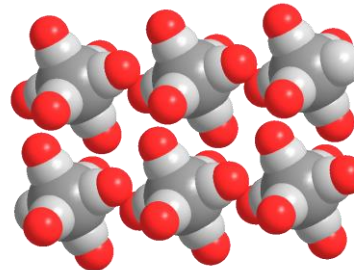
[210]

a=6.53; b=6.49 Å; φ=82.7

side



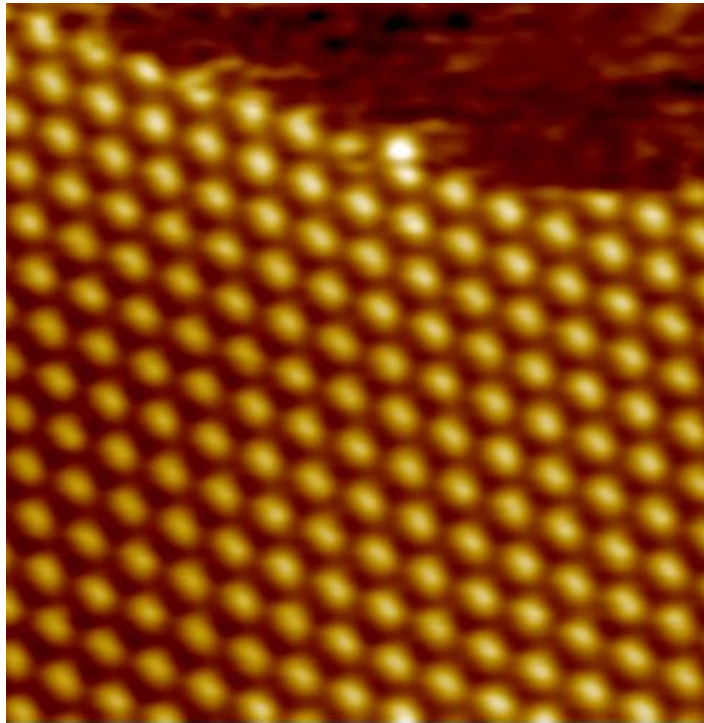
top



T.C.W.Mak,
*Zeitschrift für
Kristallographie*
166 (1984) 277-281

Arrangement is guided by forces between molecules

Mo(CO)₆ monolayer on Cu(111)



8x8 nm²; I_t= 10 pA; V_b= 20 mV

Hexagonal a=6.8 Å (±0.1)

Cu(111)+(√7x√7)R19 Mo(CO)₆

Basic 2D lattice/group : [14] Hexagonal p3m1

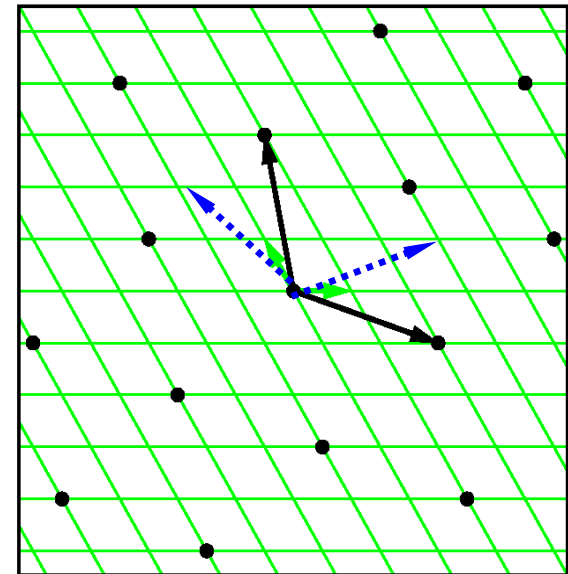
a = 2.55000, b = 2.55000, phi = 120.00

Superlattice/group : (None) Matrix = (2, -1 | 1, 3) [1]

a_s = 6.74667, b_s = 6.74667, phi_s = 120.00, 2 uni

Channel plate distortion = 0.0000

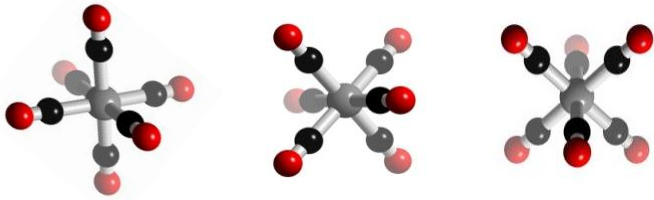
Real lattice



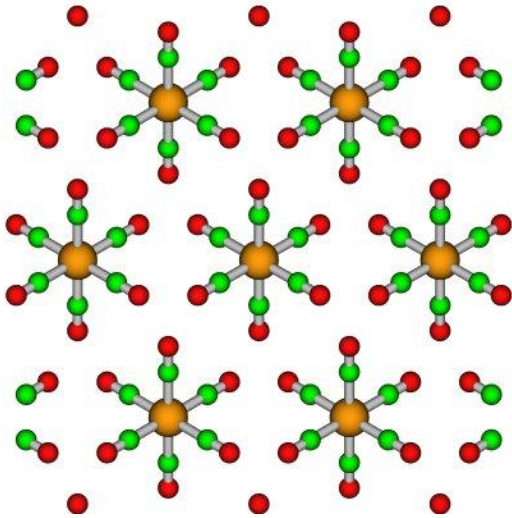
Nucleation site and molecule orientation ?

DFT calculations

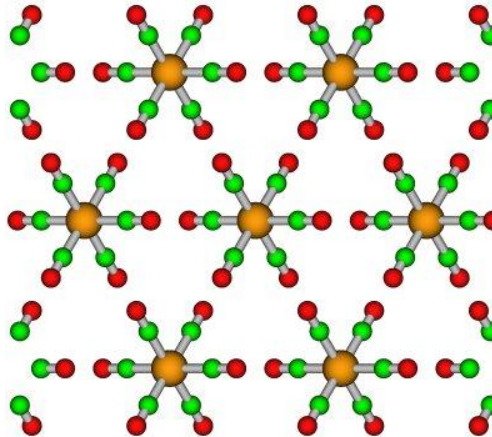
Semiempirical GGA-type density functional constructed with a long-range dispersion correction (Van-der-Waals)



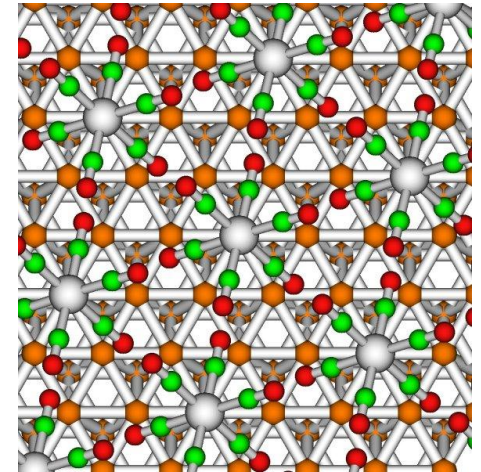
3-leg orientation



“short” O-O distances are avoided



Bulk $\text{Mo}(\text{CO})_6$:
 $E_{\text{cohesive}} = -0.58 \text{ eV/molecule}$

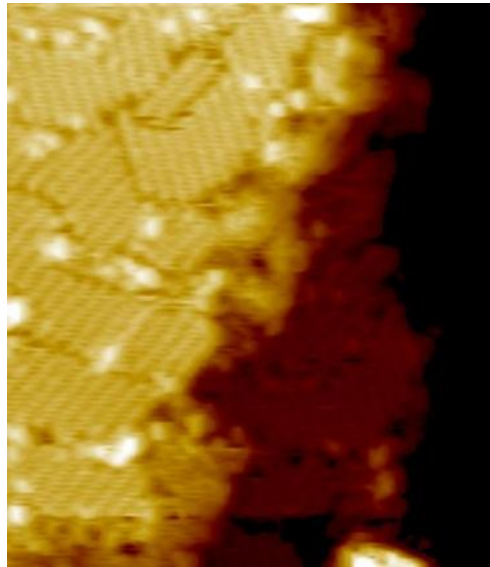


close packed (6.7 \AA)
on-top nucleation site

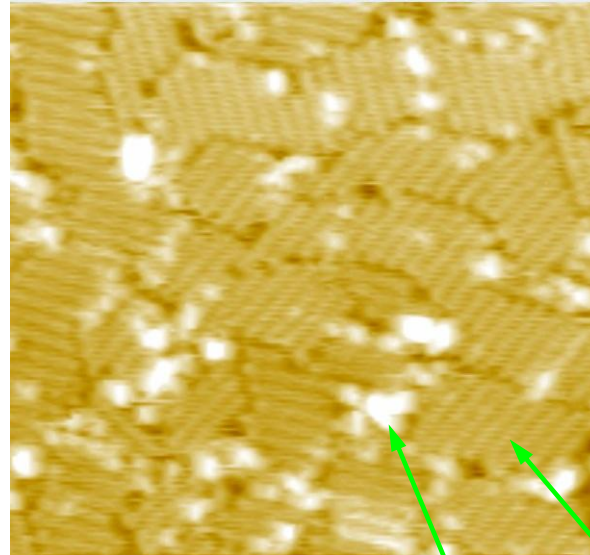
$E_{\text{ML}} = -0.70 \text{ eV/molecule}$

$\text{Cu}(111) + (\sqrt{7} \times \sqrt{7})\text{R}19 \text{ Mo}(\text{CO})_6$ confirmed by calculation

Mo(CO)₆ monolayer decomposition only on Cu(111) !



$I_t = 10 \text{ pA}; V_b = 50 \text{ mV}$



Mo clusters

Adsorbed CO domains

Chemistry ?

unstable dI/dV

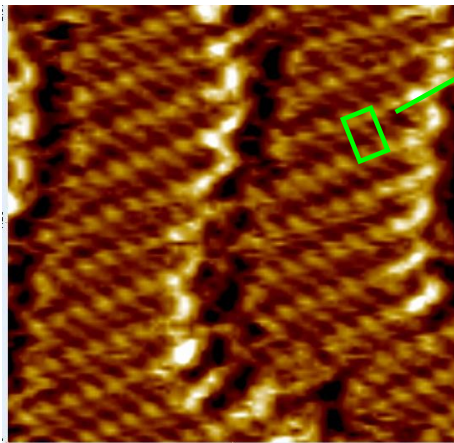
Clusters area = 10 %

($\sqrt{7} \times \sqrt{7}$)

1 ML coverage :

Mo atoms 14 %

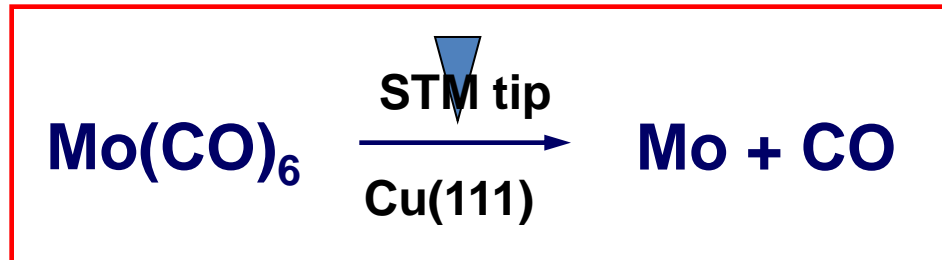
CO molecules 86 %



$2.6 \times 4.4 \text{ \AA}^2$

Cu(111) (1x2)CO !

5x5 nm; $I_t = 0.12 \text{ nA}; V_b = 100 \text{ mV}$



Conclusions

- **Monolayer adsorption of $\text{Mo}(\text{CO})_6$ was studied by STM and DFT calculations**
- **STM tip generate decomposition of $\text{Mo}(\text{CO})_6$ monolayer on Cu surface.**

Perspectives

- **Controlled decomposition conditions; nanostructures**
- **Scanning tunneling spectroscopy**
- **Multi layers growth**

Acknowledgements

Pierre Pauffert *PhD Student, Université de Bourgogne, Dijon*

ANR Project :

“Electronic Lithography through Self Assembly”

Mo(CO)₆ monolayer on Cu(111)

